

Conformational cooling and conformation selective aggregation in dimethyl sulfite isolated in solid rare gases

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Abstract

Dimethyl sulfite has three conformers of low energy, GG, GT and GG', which have significant populations in the gas phase at room temperature. According to theoretical predictions, the GT and GG' conformers are higher in energy than the GG conformer by 0.83 and 1.18 kJ mol⁻¹, respectively, while the barriers associated with the GG'→GT and GT→GG isomerizations are 1.90 and 9.64 kJ mol⁻¹, respectively. Experimental data obtained for the compound isolated in solid argon, krypton and xenon demonstrated that the GG'→GT energy barrier is low enough to allow an extensive conversion of the GG' form into the GT conformer during deposition of the matrices, the extent of the conversion increasing along the series Ar < Kr < Xe. Indeed, for substrate temperatures lower than ca. 30 K, the three conformers could be trapped in both argon and krypton matrices, but, at a given temperature, the amount of GG' form trapped in krypton is considerably smaller than in argon, while the amount of GT form increases in relation to the most stable GG form. In addition, when xenon is used, no bands due to GG' are observed in the as-deposited spectra ($T_{\text{substrate}} \geq 10$ K, the minimum substrate temperature accessible to our experimental set up), indicating that when the best relaxant gas is used the GG'→GT conversion during deposition of the matrix is complete even at 10 K. Annealing of the argon and krypton matrices shows that the increase of the temperature of the matrix first promotes the GG'→GT isomerization, and only at higher temperatures the GT→GG conversion starts to occur, in consonance with the relative energy barriers associated with these two processes. The results also indicate that dimethyl sulfite exhibits conformation selective aggregation, with the most stable form, which has the highest dipole moment, aggregating more easily than the remaining experimentally relevant conformers (GT and GG').

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1. Introduction

Dimethyl sulfite (DMSO₃) is the simplest member of the organic sulfites' family and has been shown to have many important biological applications (e.g. as insect repellent and odorant), being present, for instance, in male preputial glands [1,2]. Due to its reactivity towards different nucleofiles, DMSO₃ has also been currently used in organic synthesis [3].

The study of the vibrational spectra of the compound in different experimental conditions has been carried out by different authors [4–10]. Klaeboe [7] was the first to undertake a tentative assignment of the fundamental vibrations of liquid DMSO₃ (at room temperature) by infrared and Raman

spectroscopies. In spite of the fact that Klaeboe has not explicitly considered conformational rotamerism, this possibility was not excluded. In a more recent study of DMSO₃, Odeurs et al. [8] have investigated the gas phase conformational preferences of this molecule, based on the analysis of the observed ν S=O infrared band profile. According to that work, two conformers (forms GG and GT, shown in Fig. 1), should constitute the vast majority of the species present in the conformational equilibrium in the gaseous phase. In 1987, Barnes and Van der Veken [9] were able to isolate DMSO₃ and its perdeuterated derivative in argon and nitrogen matrices at 20 K. This was the first work to report the co-existence of three conformers of DMSO₃. The annealing of the matrices up to ca. 30 K induced the disappearance of the bands corresponding to the less stable conformers, thus giving clear evidence that the relevant barriers for conformational isomerization should be smaller than ca. 8 kJ mol⁻¹ [9,11]. In spite of the fact that the powerful methods of calculation used routinely nowadays to support the assignment of vibrational spectra were not

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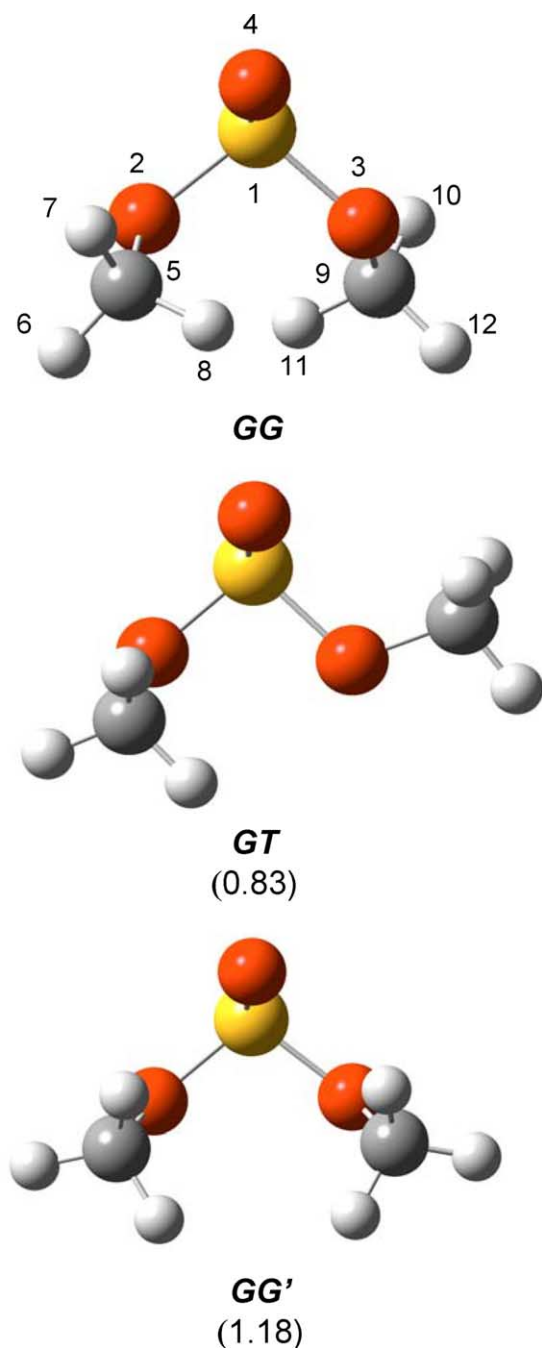


Fig. 1. Experimentally relevant conformers of DMSO₃ optimized at the DFT/aug-cc-pVDZ level of calculation with atom numbering scheme. Relative energies (kJ mol⁻¹) to the most stable conformer (GG) are shown in parentheses. The calculated O4=S1–O2–C5 and O4=S1–O3–C9 dihedral angles, which define the conformation, are 73.2°/70.8°, 68.5°/–173.2° and 85.7°/–85.7°, respectively.

available at that time, Barnes and Van der Veken were able to present a first assignment of the bands corresponding to each conformer taking advantage of isotopic labeling and carefully executed annealing experiments in both argon and nitrogen matrices. In consonance with the results of Barnes and Van der Veken [9], we have recently proved the presence of three conformers (GG, GT and GG'; Fig. 1) of DMSO₃ in argon matrices [10]. In our previous study, we have also reported

results of high-level theoretical calculations on the studied compound, using the extended quadruple- ζ aug-cc-pQVZ basis set, together with the DFT/B3LYP method. The calculated energies for the GT and GG' conformers were predicted to be 0.83 and 1.18 kJ mol⁻¹, respectively, above that of the most stable GG conformer [10]. The energy barriers associated with the GG' \rightarrow GT and GT \rightarrow GG isomerization processes were also estimated theoretically. The first was predicted to be very small (1.90 kJ mol⁻¹), while the second was found to be somewhat higher, 9.64 kJ mol⁻¹, but still low enough to eventually allow for the isomerization process to be observed during matrix deposition or by annealing of the matrix after deposition. On the other hand, the GG' \rightarrow GG process has been found to have a considerably higher energy barrier (19.5 kJ mol⁻¹) [10].

Conformational cooling (change of population towards lower energy conformational levels, but without requiring the existence of a thermodynamic temperature) has been observed to take place during matrix deposition for other compounds having different conformational states of similar energies separated by low energy barriers, such as cyanoacetic acid, methyl cyanoacetate, glycine and dimethylglycine methyl ester [12–18], for example. The observation of only two conformers (GT and GG') of DMSO₃ isolated in solid xenon [10] was a clear indication that this effect does also take place in the case of this compound. In our previous study on DMSO₃ [10], this topic was addressed briefly. In the present paper, it is investigated in detail. To fulfill this aim, DMSO₃ was isolated using matrix gases with known different relaxant capabilities (argon < krypton < xenon [13]) and a wide range of deposition temperatures.

The use of the higher temperature matrices (in particular, xenon matrices) also enabled us to investigate the aggregation process in DMSO₃. As it will be shown below, the most stable conformer of DMSO₃ (GG) seems to have a trend to aggregate easier than the higher energy GT and GG' forms.

2. Materials and methods

Dimethyl sulfite was obtained from Aldrich (purity 99%). The IR spectra were collected, with 0.5 cm⁻¹ spectral resolution, on a Mattson (Infinity 60AR Series) fourier transform infrared spectrometer, equipped with a deuterated triglycine sulfate (DTGS) detector and a Ge/KBr beamsplitter. Necessary modifications of the sample compartment of the spectrometer were made in order to accommodate the cryostat head and allow efficient purging of the instrument by a stream of dry N₂ to remove water and CO₂ vapors.

In the matrix isolation experiments, a glass vacuum system and standard manometric procedures were used to deposit the matrix gas (argon N60, krypton N48 or xenon N48, all of them obtained from Air Liquide). Matrices were prepared by co-deposition, onto the cooled CsI substrate of the cryostat, of the matrix gas and DMSO₃, which was sublimated using a specially designed doubly thermostable Knudsen cell with shut-off possibility. The container of the Knudsen cell was kept at low temperature (ca. –100 °C) to reduce the vapor pressure of the compound and enable a better control of the amount of

DMSO₃ being deposited, while the nozzle was kept at room temperature (298 K). All experiments were done on the basis of an APD Cryogenics close-cycle helium refrigeration system with a DE-202A expander. Different deposition temperatures were used, ranging from 10 to 30 K (Ar and Kr) or 10 to 50 K (Xe). After depositing of the compound, annealing experiments were performed until a temperature of 35 K (Ar), 55 K (Kr) or 60 K (Xe) was attained.

3. Results and discussion

The as-deposited spectra of DMSO₃ in argon, krypton and xenon matrices, using a substrate temperature of 10 K (with the valve nozzle at room temperature, 298 K), are displayed in Fig. 2. The DFT (B3LYP)/aug-cc-pVQZ calculated spectra [10] for the three experimentally relevant conformers of DMSO₃ (GG, GT and GG') are also shown in this figure for comparison. Tables 1–3 show the assignments made for the spectra collected in the three types of matrices. As it was mentioned in our previous work [10], the matrix isolation spectra of DMSO₃, as found for other analogous sulfur containing compounds [19], exhibit band-splitting (which in DMSO₃ becomes more evident for the features assigned to the GT and GG' forms) due to occupancy of different matrix-sites. This effect is more pronounced in argon and krypton matrices, but it also exists in xenon. Some of the bands originated in different matrix-sites displayed intensity reorganization when the compound was deposited at different temperatures of the substrate, indicating that the stability of the different matrix-sites shall be at least slightly different.

As shown in Fig. 2, using $T_{\text{substrate}} = 10$ K the GG, GT and GG' conformers could be trapped in argon and krypton matrices, but not in xenon, where no bands of GG' were observed (this may be particularly well seen in the $\nu_{\text{S=O}}$ stretching region—Fig. 2a, where the bands due to the different conformers appear well separated, but is also consistently observed along the spectra—see, for example, the 1020–950 and 760–650 cm^{-1} regions shown in Fig. 2b and c). As mentioned before, this fact indicates the occurrence of extensive conformational cooling during deposition of the xenon matrix, due to isomerization of conformer GG' into form GT. Accordingly to this interpretation, the relative intensity of the bands corresponding to GT in the Xe matrix should be nearly equal to the sum of the relative intensities of the bands of GT and GG' in the other two matrices. On the other hand, the relative intensity of the bands due to conformer GG should be nearly equal in the three matrices, since no isomerization from GT and GG' into GG is expected to take place at this low temperature of deposition due to the higher energy barriers associated with both the GT → GG and GG' → GG processes (9.4 and 19.5 kJ mol^{-1} , respectively).

The relative populations of the three DMSO₃ conformers in the three studied matrices deposited at 10 K are presented in the graphic shown in Fig. 3. The populations were estimated from the integral intensity of the bands in the 1250–1180 cm^{-1} spectral region ($\nu_{\text{S=O}}$), weighted by the corresponding calculated [DFT(B3LYP)/aug-cc-pVQZ] intensities. In consonance with the expectations, the population of the GG form trapped in Ar and Kr matrices and the sum of the populations of GT and GG' forms are approximately the same.

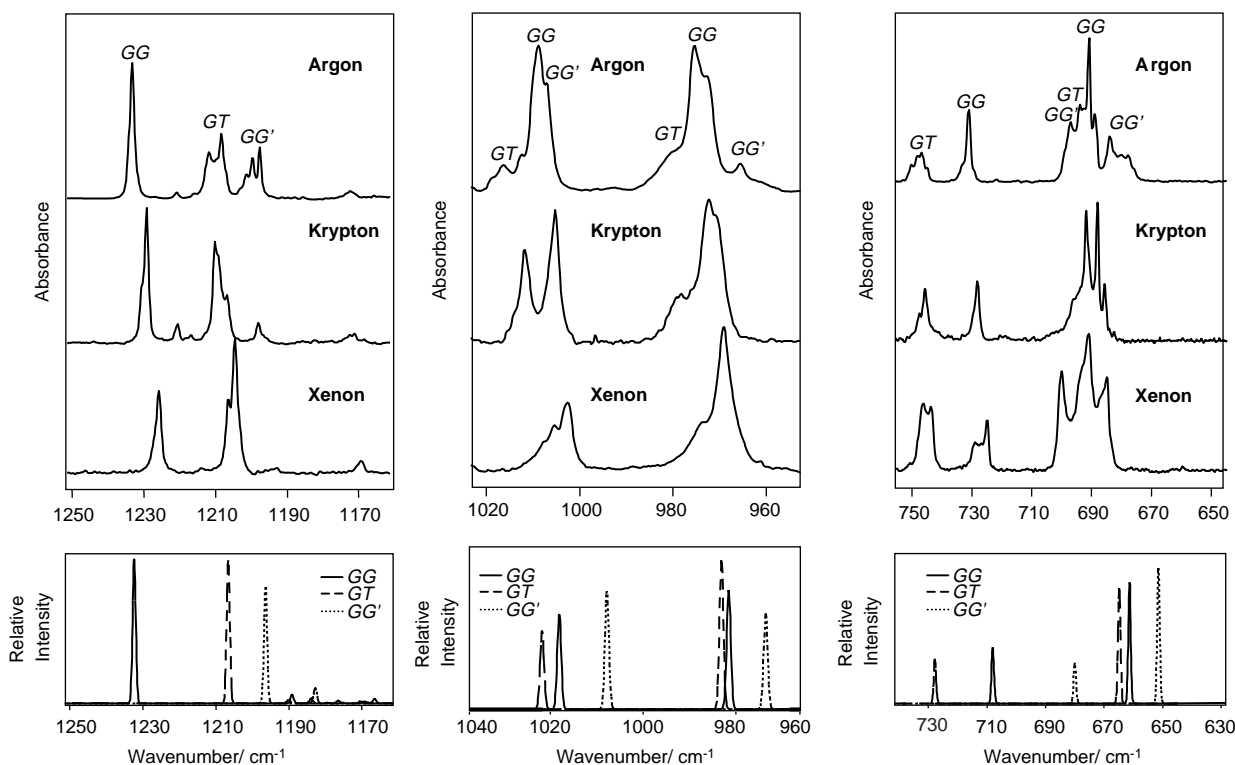


Fig. 2. Infrared spectra of DMSO₃ in argon, krypton and xenon matrices obtained immediately after deposition at 10 K. Calculated spectra for the GG, GT and GG' conformers are also displayed for comparison: (a) 1240–1185 cm^{-1} region; (b) 1020–960 cm^{-1} region; (c) 755–660 cm^{-1} region.

Table 1
Observed frequencies (cm^{-1}) for the GG form of dimethyl sulfite monomer in argon, krypton and xenon matrices

Approximate description	Calculated frequency	Calculated intensity	Observed frequency Ar (10 K)		Observed frequency Kr (10 K)		Observed frequency (10 K)	
			More stable site	Less stable sites	More stable site	Less stable sites	More stable sites	Less stable sites
$\nu(\text{C9-H}_3)'_{\text{as}}$	3147.2	6.1	3028.0		3017.0		3022.3	
$\nu(\text{C5-H}_3)'_{\text{as}}$	3133.7	9.4	3028.0		3017.0		3022.3	
$\nu(\text{C9-H}_3)''_{\text{as}}$	3124.6	14.6	3014.5		3004.5		3009.6	
$\nu(\text{C5-H}_3)''_{\text{as}}$	3089.4	22.8	3005.1		2994.9		3003.9	
$\nu(\text{C9-H}_3)_s$	3048.0	30.4	2960.5		2945.3		2969.5/2945.4	
$\nu(\text{C5-H}_3)_s$	3024.2	41.6	2955.1		2939.0		2935.2/2928.8	
$\delta(\text{C9-H}_3)'_{\text{as}}$	1509.5	4.8	1469.1		1466.1		1462.4	
$\delta(\text{C5-H}_3)''_{\text{as}}$	1504.8	13.4	1464.5		1461.5		1458.5	
$\delta(\text{C5-H}_3)'_{\text{as}}$	1492.0	9.3	1455.7		1453.7		1449.9	
$\delta(\text{C9-H}_3)''_{\text{as}}$	1485.7	7.9	1451.3		1451.7		1445.7	
$\delta(\text{C5-H}_3)_s$	1467.9	0.3	1446.8		1448.7		1442.9	
$\delta(\text{C9-H}_3)_s$	1458.7	0.3	1428.9		1426.0		Not observed	
$\nu(\text{S=O})$	1231.7	151.6	1233.2	1231.8	1230.6	1229.2	1227.5	1225.9
$\gamma(\text{C5-H}_3)'$	1188.5	9.3	1185.4		1185.2			
$\gamma(\text{C9-H}_3)'$	1178.8	0.7	Not observed		Not observed			
$\gamma(\text{C5-H}_3)''$	1174.6	0.7	Not observed		Not observed			
$\gamma(\text{C9-H}_3)''$	1169.8	2.2	1165.7		1165.5			
$\nu(\text{C5-O})$	1017.5	59.1	1009.7	1009.2	1005.4			
$\nu(\text{C9-O})$	981.1	99.2	972.1	975.6	972.3	971.0	969.2	
$\gamma(\text{S=O})$	707.7	10.5	733.2	730.8	728.0	729.8	724.7	728.8/727.1
$\nu(\text{S-O2})$	661.1	36.6	693.8/688.5	692.1/690.7/689.1	687.1	691.7	684.8	690.8
$\nu(\text{S-O3})$	563.3	7.0	578.2		576.1		580.0	
$\delta(\text{OSO})$	427.0	1.1	Not observed		Not observed		Not observed	
$\delta(\text{S=O})$	401.5	7.4						
$\delta(\text{S-O3-C9})$	250.3	12.0						
$\delta(\text{S-O2-C5})$	241.8	1.1						
$\tau(\text{S-O3})$	138.3	4.1	Not investigated		Not investigated		Not investigated	
$\tau(\text{C5-H}_3)$	91.3	1.0						
$\tau(\text{C9-H}_3)$	84.6	0.3						
$\tau(\text{S-O2})$	77.3	2.4						

B3LYP/aug-cc-pVQZ calculated frequencies and intensities (km mol^{-1}) [10] are given for comparison; ν , bond stretching; δ , bending; γ , rocking; τ , torsion; s, symmetric; as, asymmetric. See Ref. [10] for definition of internal coordinates and Fig. 1 for atom numbering.

In addition, the relative population of GT increases relatively to that of GG' in going from Ar to Kr (as already mentioned, in Xe, no GG' bands are observed). These results unequivocally prove that the GG' \rightarrow GT isomerization took place during deposition of the various matrices at 10 K, decreasing in importance from Xe (where it occurs completely) to Ar. There are, however, some additional conclusions that can be extracted from the data shown in Fig. 3, that are subtler:

The relative populations of the conformers at 298 K, estimated from the theoretical calculations (assuming the Boltzmann distribution and using the DFT(B3LYP)/aug-cc-pVQZ calculated relative conformational energies) are GG: 48.8%, GT + GG': 35.0 + 15.1 = 50.1%. On the other hand, the experimental values obtained for the argon matrix are GG: 36%, GT + GG': 37 + 27 = 64%. In krypton, the corresponding values are GG: 42%, GT + GG': 54 + 4 = 58%. Hence, the results seem to indicate that the calculations slightly overestimate the energy of at least one of the higher energy conformers (GT, GG') comparatively to GG (the sum of the calculated populations of GT and GG' is consistently smaller than the experimental values). In addition, the results also point to occurrence in small amount of GT \rightarrow GG isomerization

during deposition of the Kr matrix (as already mentioned, Kr is a more relaxant gas than Ar). On the other hand, with all probability, this process does not occur during deposition of argon matrix at 10 K.

In the 10 K as-deposited xenon matrix, the population of the most stable GG form, though comparable to those observed in Ar and Kr matrices, was found to be slightly smaller than in these two latter matrices (30 vs. 36% in argon and 42% in krypton). At first, this result appeared somewhat surprising, since it could be expected the population of this form to increase slightly relatively to what was found in Kr (xenon being a better relaxant than Kr could eventually led to increase the possibility of occurrence of the GT \rightarrow GG process during deposition of the Xe matrix when compared to Kr). An explanation for this observation could only be obtained after recording of spectra in the various matrices using higher substrate temperatures during deposition. Fig. 4 shows the relative conformational populations estimated as previously from data collected for a substrate temperature of 30 K in matrices of the three support-gases used. As it can be seen from this figure, according to the expectations, at this substrate temperature both GG' \rightarrow GT and GT \rightarrow GG processes occur

Table 2
Observed frequencies (cm^{-1}) for the GT form of dimethyl sulfite monomer in argon, krypton and xenon matrices

Approximate description	Calculated frequency	Calculated intensity	Observed frequency Ar (10 K)		Observed frequency Kr (10 K)		Observed Frequency Xe (10 K)	
			More stable site	Less stable sites	More stable site	Less stable sites	More stable site	Less stable sites
$\nu(\text{C5-H}_3)'_{\text{as}}$	3150.4	7.3	3028.0		3017.0		3022.3	
$\nu(\text{C5-H}_3)''_{\text{as}}$	3135.1	8.8	3028.0		3017.0		3022.3	
$\nu(\text{C9-H}_3)'_{\text{as}}$	3134.1	10.0	3028.0		3017.0		3022.3	
$\nu(\text{C9-H}_3)''_{\text{as}}$	3095.4	19.4	3005.1		2994.9		3003.9	
$\nu(\text{C5-H}_3)_s$	3055.2	30.2	2960.5		2945.3		2969.5/2945.4	
$\nu(\text{C9-H}_3)_s$	3020.8	49.3	2952.8		2939.0		2935.2/2928.8	
$\delta(\text{C9-H}_3)'_{\text{as}}$	1508.1	14.4	1469.1		1466.1		1462.4	
$\delta(\text{C5-H}_3)''_{\text{as}}$	1507.3	6.8	1467.8		1463.6		1458.5	
$\delta(\text{C9-H}_3)''_{\text{as}}$	1492.6	7.5	1457.6		1456.6		1449.9	
$\delta(\text{C5-H}_3)'_{\text{as}}$	1489.5	7.9	1454.0		1452.6		1445.7	
$\delta(\text{C9-H}_3)_s$	1468.8	0.3	1446.8		1448.7		1442.9	
$\delta(\text{C5-H}_3)_s$	1459.0	0.4	1428.9		1426.0		Not observed	
$\nu(\text{S=O})$	1206.0	150.0	1208.6/1207.0	1211.7/1211.2 1209.7	1210.1	1209.1/1206.8	1203.3	1206.5/1204.5
$\gamma(\text{C9-H}_3)'$	1189.5	3.3	1189.6		1189.2		1194.4	
$\gamma(\text{C5-H}_3)'$	1183.3	5.1	1187.9		1186.7		1185.8	1182.6
$\gamma(\text{C9-H}_3)''$	1175.9	2.8	Not observed		Not observed		Not observed	
$\gamma(\text{C5-H}_3)''$	1168.7	1.9	Not observed		Not observed		Not observed	
$\nu(\text{C9-O})$	1021.2	131.5	1016.6	1018.8/1012.3	1011.8	1015.7/1014.2	1005.4	1009.9/1007.6
$\nu(\text{C5-O})$	982.7	249.1	980.1		979.3/978.3		973.3	
$\nu(\text{S-O2})$	727.5	86.2	748.3/747.3	750.1/746.9	745.5	747.4	746.4	747.5/744.9
$\nu(\text{S-O3})$	664.6	225.3	696.9	Not observed	696.2	693.7	699.6	692.7
$\gamma(\text{S=O})$	566.1	15.5	578.2		576.1		580.0	
$\delta(\text{S=O})$	456.8	2.5	Not observed		Not observed		Not observed	
$\delta(\text{OSO})$	383.8	5.7						
$\delta(\text{S-O2-C5})$	278.6	4.9						
$\delta(\text{S-O3-C9})$	218.8	6.4						
$\tau(\text{C9-H}_3)$	141.6	4.2	Not investigated		Not investigated		Not investigated	
$\tau(\text{C5-H}_3)$	121.7	2.5						
$\tau(\text{S-O2})$	93.2	4.3						
$\tau(\text{S-O3})$	72.8	3.4						

B3LYP/aug-cc-pVQZ calculated frequencies and intensities (km mol^{-1}) [10] are given for comparison; ν , bond stretching; δ , bending; γ , rocking; τ , torsion; s, symmetric; as, asymmetric. See Ref. [10] for definition of internal coordinates and Fig. 1 for atom numbering.

extensively during deposition of the Ar and Kr matrices and the population of the most stable GG form is 100% in the Kr matrix and nearly this value in the less relaxant Ar matrix. However, the situation was found to be dramatically different in the xenon matrix, with the GT/GG population ratio being approximately the same as obtained for matrices deposited at 10 K (and also other temperatures between 10 and 30 K). Such behavior cannot be explained in terms of conformational cooling, since the GG form is the most stable conformer in the matrix. Indeed, an inversion in the relative stability of the GG and GT conformers in this matrix, when compared to the other matrices (and also to the gaseous phase), looks very improbable, since GT is the less polar conformer and shall in fact be somewhat destabilized in the most polarizable xenon matrix (calculated dipole moments are 3.91, 2.00 and 2.67 debye for GG, GT and GG', respectively [10]). The observations can however be easily explained assuming that aggregation is taking place during deposition of the xenon matrices, with the most polar GG form showing a greater trend to associate than the remaining (less polar) conformers. This effect would then mask the effect of conformational cooling during deposition of the Xe matrices, i.e. because the GG form

aggregates easier than the GT form, its population reduces relatively to that of the latter conformer. Naturally, this effect must be more pronounced for higher substrate temperatures and, as result of simultaneous occurrence of the two processes (conformational cooling, leading to increase the population of GG, and conformationally selective aggregation, leading to reduce it relatively to that of GT), the GT/GG population ratio stays nearly invariant within the temperature range covered by our experiments. The fact that the GG population observed in the as-deposited 10 K xenon matrix is slightly smaller than in the as-deposited 10 K krypton and argon matrices is then an indication that even at this temperature some aggregation of the compound is taking place. Very unfortunately, in xenon matrices aggregates usually give rise to very broad bands, which most of times are not possible to discriminate from the baseline [20]. For a relatively small amount of aggregates, as expected in the present case for the Xe matrices deposited at the lowest substrate temperatures, observation of these bands would be highly improbable. Formation of aggregates is, however, quite clear in Xe matrices for higher temperatures of deposition (Fig. 5), and, symptomatically, systematically more important, at a given temperature, than in both Ar and Kr

Table 3

Observed frequencies (cm^{-1}) for the GG' form of dimethyl sulfite monomer in argon and krypton matrices. B3LYP/aug-cc-pVQZ calculated frequencies and intensities (km mol^{-1}) [10] are given for comparison

Approximate description	Calculated frequency	Calculated intensity	Observed frequency Ar (10 K)		Observed Frequency Kr (10 K)	
			More stable site	Less stable sites	More stable site	Less stable sites
$\nu(\text{CH}_3)'_{\text{as}}s$	3138.0	3.4	3028.0		3017.0	
$\nu(\text{CH}_3)'_{\text{as}} \text{as}$	3137.3	8.7	3028.0		3017.0	
$\nu(\text{CH}_3)''_{\text{as}} s$	3116.3	12.8	2986.0		2978.8	
$\nu(\text{CH}_3)''_{\text{as}} \text{as}$	3113.5	25.1	2986.0		2978.8	
$\nu(\text{CH}_3)_s s$	3041.4	62.3	2960.5		2959.5	
$\nu(\text{CH}_3)_s \text{as}$	3037.5	9.4	2960.5		2959.5	
$\delta(\text{CH}_3)'_{\text{as}} s$	1508.9	2.9	1469.1		1466.1	
$\delta(\text{CH}_3)''_{\text{as}} s$	1501.5	9.4	1460.5		1458.9	
$\delta(\text{CH}_3)'_{\text{as}} \text{as}$	1500.1	16.3	1460.5		1458.9	
$\delta(\text{CH}_3)''_{\text{as}} \text{as}$	1491.1	2.9	1455.7		1453.7	
$\delta(\text{CH}_3)_s s$	1461.9	2.1	1428.9		1426.0	
$\delta(\text{CH}_3)_s \text{as}$	1454.8	0.3	1422.7		1420.7	
$\nu(\text{S}=\text{O})$	1195.7	122.3	1197.7	1202.9/1199.7	1198.0	1196.6/1195.6
$\gamma(\text{CH}_3)'_{\text{as}}$	1183.3	0.9	Not observed		Not observed	
$\gamma(\text{CH}_3)'_s$	1182.1	16.4	1180.4		1181.3	
$\gamma(\text{CH}_3)''_{\text{as}}$	1167.6	0.9	Not observed		Not observed	
$\gamma(\text{CH}_3)''_s$	1165.9	4.9	1165.8		1165.5	
$\nu(\text{C}-\text{O})_s$	1007.3	197.5	1006.9		1001.7	
$\nu(\text{C}-\text{O})''_{\text{as}}$	973.2	160.0	965.7		971.0	
$\nu(\text{S}-\text{O})_s$	679.7	79.8	698.3	699.1	698.2	700.0
$\nu(\text{S}-\text{O})''_{\text{as}}$	651.2	266.7	682.2	685.0/683.7/680.1/ 677.5	682.3	685.5
$\gamma(\text{S}=\text{O})$	599.2	15.3	615.3		616.4	
$\delta(\text{S}=\text{O})$	479.3	0.0	Not observed		Not observed	
$\delta(\text{OSO})$	354.1	7.6				
$\delta(\text{SOC})_{\text{as}}$	273.5	12.5				
$\delta(\text{SOC})_s$	266.1	11.4				
$\tau(\text{CH}_3)_{\text{as}}$	144.2	1.8	Not investigated		Not investigated	
$\tau(\text{S}-\text{O})_{\text{as}}$	93.9	2.2				
$\tau(\text{CH}_3)_s$	84.3	0.0				
$\tau(\text{S}-\text{O})_s$	44.2	0.7				

ν , Bond stretching; δ , bending; γ , rocking; τ , torsion; s , symmetric; as , asymmetric. See Ref. [10] for definition of internal coordinates and Fig. 1 for atom numbering.

matrices. It is worth noting that conformation selective aggregation has been found to occur in different compounds, and matrix isolation infrared spectroscopy has been shown to constitute a powerful technique to probe such phenomenon [15,20].

In our previous study on DMSO_3 [10], we had already shown that annealing of the argon matrix deposited at 15 K to higher temperatures led to spectral changes which are consistent with occurrence of $\text{GG}' \rightarrow \text{GT}$ isomerization within the temperature range 15–21 K while the $\text{GT} \rightarrow \text{GG}$ processes started to occur significantly only for a temperature higher than 21 K. As mentioned above, the different triggering temperatures for these two processes are a direct consequence of the different energy barriers associated with them (1.90 and 9.64 kJ mol^{-1} , respectively). The consecutive nature of the isomerization processes could also now be unequivocally established for the krypton matrices, as shown by the change in the relative intensity of the $\nu\text{S}=\text{O}$ bands as a function of the substrate temperature during deposition. Fig. 6 shows the data obtained for substrate temperatures of 10, 20 and 30 K. The spectra were normalized by the band of the less stable Kr

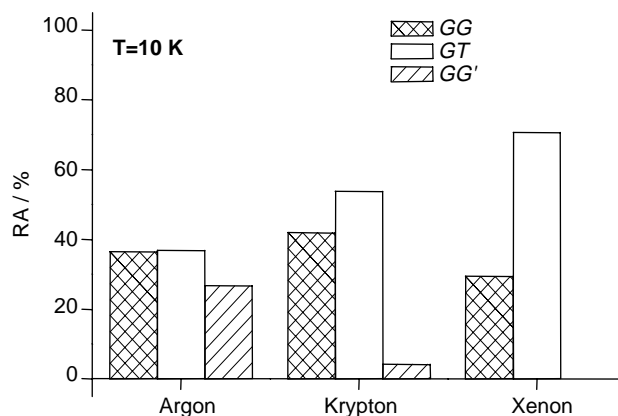


Fig. 3. Graph bar showing the experimental relative abundances (RA) of GG, GT and GG' conformers trapped in argon, krypton and xenon matrices using a substrate temperature during deposition of 10 K. The populations were estimated from the integral intensity of the bands in the 1250–1180 cm^{-1} ($\nu\text{S}=\text{O}$), weighted by the corresponding calculated [DFT(B3LYP)/aug-cc-pVQZ] intensities.

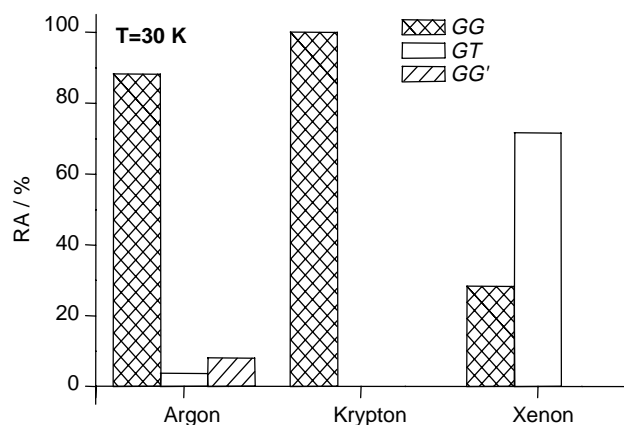


Fig. 4. Graph bar showing the experimental relative abundances (RA) of GG, GT and GG' conformers trapped in argon, krypton and xenon matrices using a substrate temperature during deposition of 30 K. The populations were estimated from the integral intensity of the bands in the 1250–1180 cm^{-1} ($\nu\text{S}=\text{O}$), weighted by the corresponding calculated [DFT(B3LYP)/aug-cc-pVQZ] intensities.

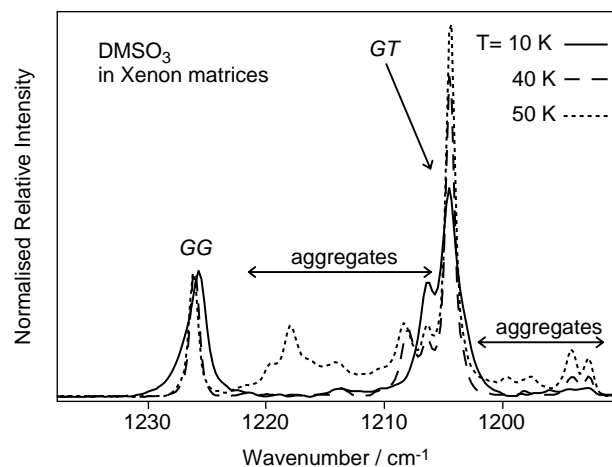


Fig. 5. The S=O stretching region of the infrared spectra of DMSO₃ isolated in xenon matrices using different substrate temperatures during deposition: 10, 40 and 50 K. The spectra were normalized to the total intensity of the features ascribed to the GG form.

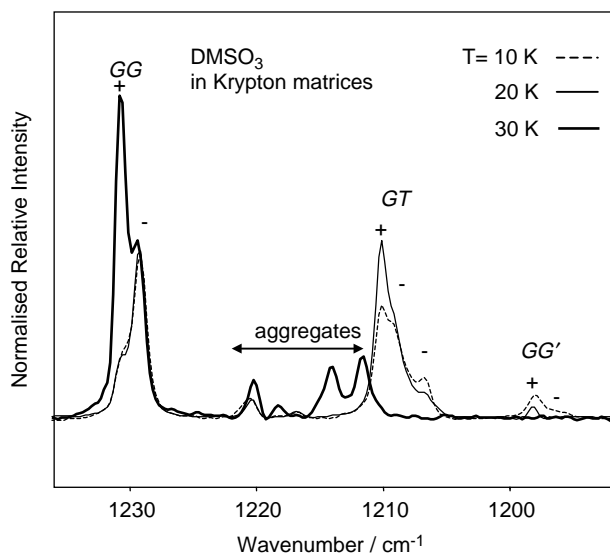


Fig. 6. The S=O stretching region of the infrared spectra of DMSO₃ isolated in krypton matrices using different substrate temperatures. The temperature of the optical substrate during deposition of the samples was kept at 10 K (dashed line), 20 K (thin solid line) and 30 K (thick solid line). The spectra were normalized to the intensity of the GG band at 1231.8 cm^{-1} . Signs (+ and -) refer to bands originated in the most stable and less stable matrix-sites, respectively (see also Tables 1–3).

matrix site of GG, at 1229.2 cm^{-1} . From this figure, it can be easily noticed that until a substrate temperature of ca. 20 K conformer GG' converts into form GT (simultaneously, inter-site conversion led to the conversion of GT Kr matrix less stable sites—giving rise to the bands at 1209.1 and 1206.8 cm^{-1} —into the most stable site—associated with the band at 1210.1 cm^{-1}). When deposition was made with substrate temperatures above ca. 20 K, conversion of GT into GG occurs, leading to disappearance of the bands due to GT and relative increase of the band of the most stable site of GG at 1230.6 cm^{-1} . It is interesting to note that the GT→GG

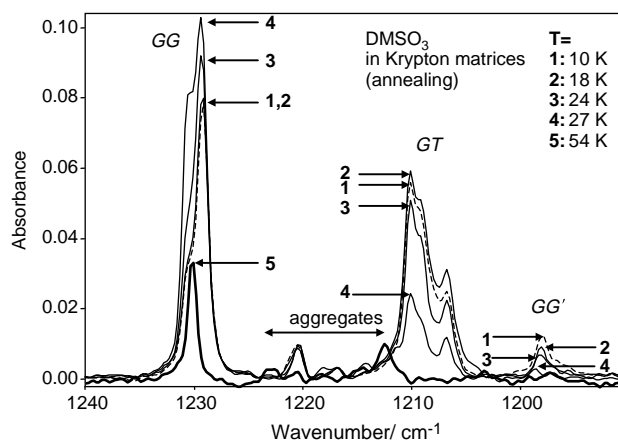


Fig. 7. The S=O stretching region of the infrared spectrum of DMSO₃ isolated in krypton matrices. The temperature of the optical substrate during deposition of the sample was kept at 10 K (dashed line—trace 1). The remaining traces (2–5) correspond to spectra obtained at different temperatures during the annealing of the matrix.

conversion leads to the production of GG form in its most stable site. The same seems to be true also for the production of GT from GG', though in this case the results are not so clear due to occurrence of important simultaneous inter-site conversion of GT (and also because the amount of GG' initially trapped in the Kr matrices is small).

As it could be anticipated, the results of the annealing of the matrices after deposition resemble very much those obtaining in the series of experiments discussed above where the deposition temperature was varied. Fig. 7 shows the results obtained in krypton, that also closely follow those previously obtained for the compound isolated in argon [10]. In the temperature range 10–18 K, essentially two phenomena are observed: GG'→GT isomerization and inter-site conversion of GT (compare traces 1 and 2 in Fig. 7). The features due to GG' and GT decrease and increase in intensity, respectively, while

those due to GG do not change. At 24 K (trace 3 in Fig. 7), the GT→GG conversion has already started and, consequently, the bands due to GG increase in intensity at expenses of those due to GT (the GG'→GT isomerization continues to occur leading to a continued decrease in intensity of the bands due to GG'). At 27 K (trace 4 in Fig. 7) the GG'→GT isomerization was practically completed and the GT→GG process has also occurred in large extension. Finally, at 54 K (trace 5), both processes were concluded while aggregation has also occurred extensively (only the most stable matrix site of the GG monomer is still somewhat populated).

4. Conclusion

Conformational cooling occurring both during and after deposition of DMSO₃ in cryogenic matrices of Ar, Kr and Xe has been analyzed. As it has been also observed for other compounds [12–18], the degree of the conformational cooling depends on the nature of the matrix gas host, being more important along the series Xe > Kr > Ar.

The low energy barrier for the GG'→GT conversion (1.90 kJ mol⁻¹) allows that extensive isomerization of the GG' into the GT conformer take place during deposition of the matrices. In xenon, GG' completely converts to GT during deposition even at a temperature as low as 10 K. On the other hand, the GT→GG process has a considerably higher energy barrier (9.64 kJ mol⁻¹) and conversion of GT into GG occurs only when deposition of the matrix is performed at higher temperatures (higher than ca. 20 K). Very interestingly, this process leads to formation of GG into its most stable matrix-site. Identical situation does also probably occur relatively to GT formed in the GG'→GT process, though the extensive inter-site conversion for GT and GG' led us to consider this conclusion only as tentative. The annealing experiments performed in both argon and krypton matrices strongly supported the results obtained in the experiments where the substrate temperature was varied, regarding the consecutive nature of the two isomerization processes investigated.

When compared with the other two types of matrices, the results obtained in xenon were not so successful in the study of the conformational isomerization processes. Firstly, because no GG' conformer could be trapped in the Xe matrices due to fast conversion to GT. Secondly, because aggregation was found to be considerably more important in this case. However, from these studies it was possible to conclude about the different ability of the conformers to aggregate. It was found that DMSO₃ exhibits conformation selective aggregation

capabilities, with the most polar GG conformer exhibiting a greater ability to aggregate than the remaining forms.

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